

Willkommen
Welcome
Bienvenue



Development of reactive joining technologies for electronic packaging and assembly

Workshop “Miniaturized Photonic Packaging”, 16.05.2017, CSEM, Alpnach

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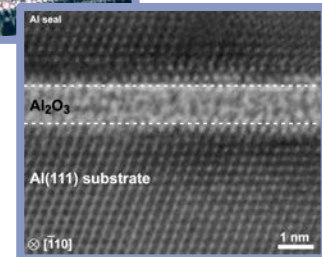
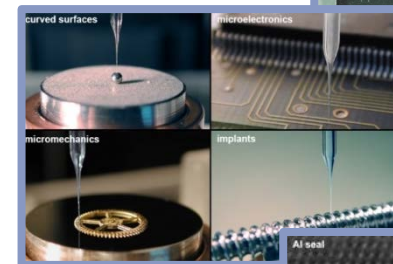
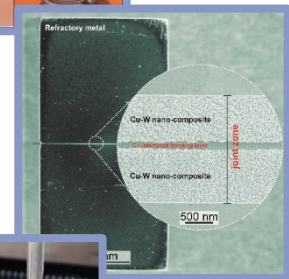
Laboratory Joining Technologies & Corrosion at Empa

Our portfolio

- Advanced Joining Technologies
(soldering, brazing, TLP, diffusion bonding, micro- & nano-joining)
- Corrosion Management
(investigations of corrosion failures, mechanisms and prevention strategies)
- Surface & Interface Engineering
(of metals, alloys, oxide films and their coating systems)

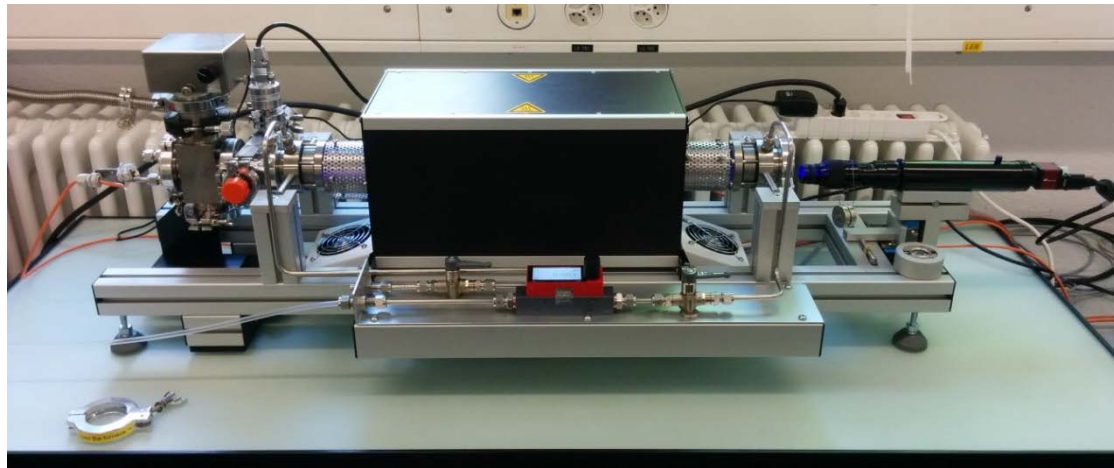
Our expertise within the Swiss Photonic Packaging Laboratory

- Custom-designed solutions in the field of joining: brazing, soldering, diffusion bonding, transient liquid phase bonding, development of nanostructured filler alloys, coatings and foils,...



Laboratory Joining Technologies & Corrosion at Empa

New wetting furnace, financial support from **SWISS PHOTONICS**



Purpose

- investigation of wettability (contact angle, spreading) under controlled conditions (t , T , atmosphere)
- generally: visual inspection of samples at high temperature under controlled atmosphere

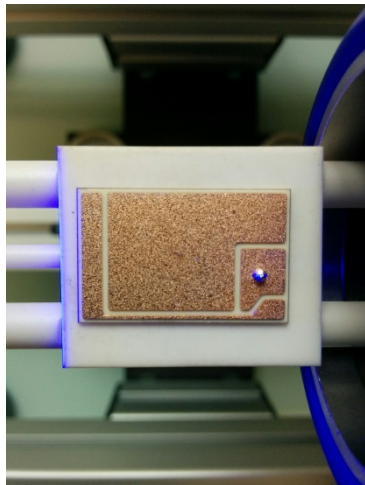
Specifications

- quartz tube furnace
 - max. heating rate: ca. 20 K/min
 - max. T : ca. 1000 °C
- atmospheres
 - controlled flow rates: inert, reducing, oxidising
 - vacuum (HV range)

Laboratory Joining Technologies & Corrosion at Empa

New wetting furnace, financial support from **SWISS PHOTONICS**

- Example: Sn pearl on DCB substrate

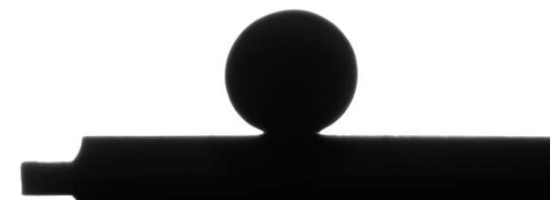
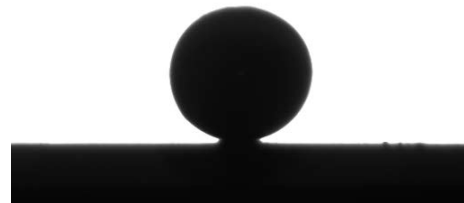


$HR = 5 \text{ K/min}$

as received

cleaned

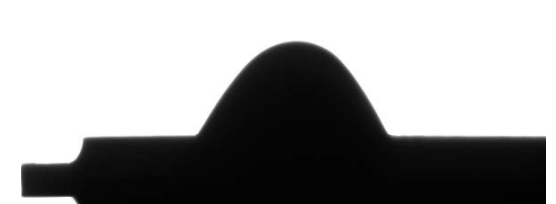
$T = 18 \text{ }^\circ\text{C},$
 $t = 0 \text{ min}$



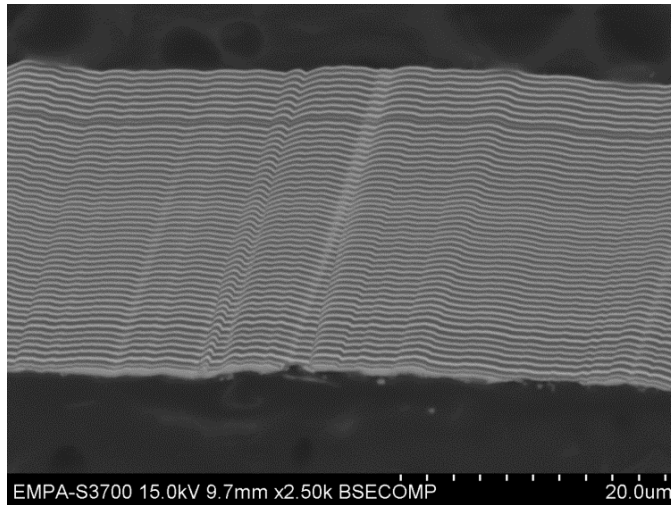
$T = 265 \text{ }^\circ\text{C},$
 $t = 49 \text{ min}$



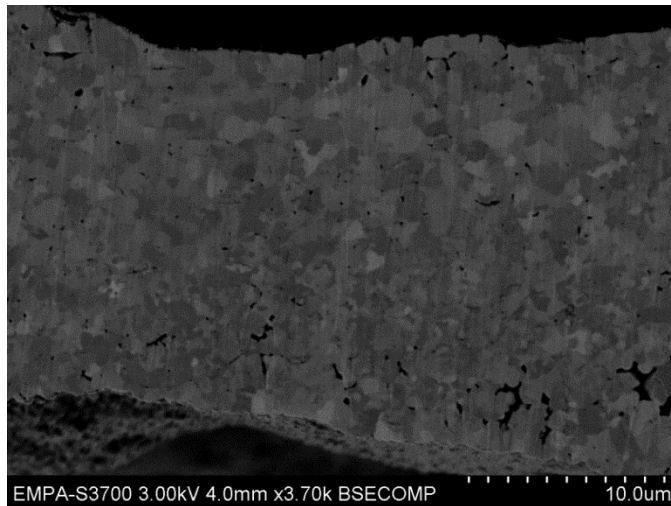
$T = 295 \text{ }^\circ\text{C},$
 $t = 55 \text{ min}$



Reactive nano-multilayers



before reaction:
nano-multilayers



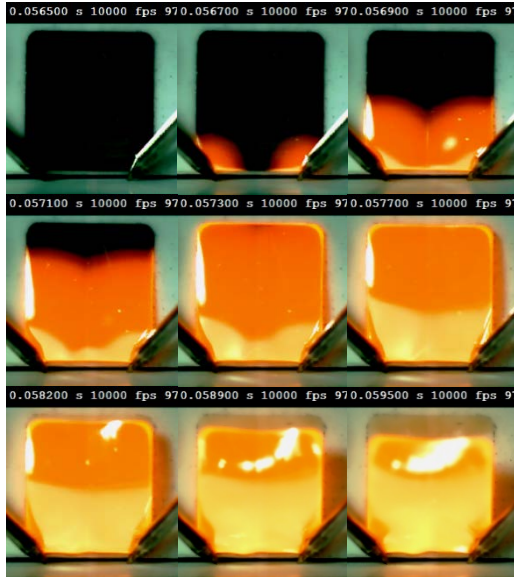
after reaction:
intermetallic
phase

cross-sections of a nano-multilayer
foil before and after reaction



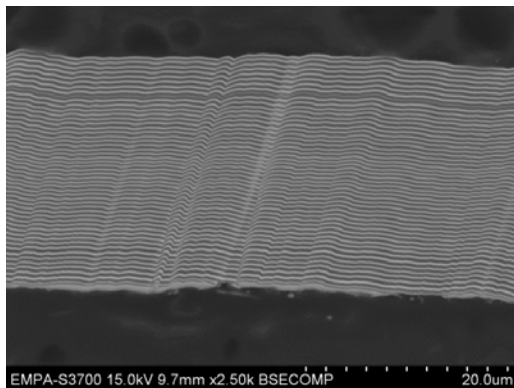
high-speed recording of a
reacting Nanofoil[®],
total time: 2.5 milliseconds

Reactive nano-multilayers



Key facts

- alternating layers of metals (e.g. Ni+Al)
- internal heat production by metal-metal reaction, no gas phase involved
- high reaction temperatures (>1000 °C)
- high reaction speeds (1-50 m/s)
- defined heat generation by variation of system and total thickness (10 - 250 μm)



type	heat release	example
low	30 - 59 kJ/mol-atom	Al/Ti
medium	60 - 89 kJ/mol-atom	Ni/Al
high	> 90 kJ/mol	Al/Pd

➔ idea: usage as internal heat source for soldering/brazing

Reactive nano-multilayers

Development

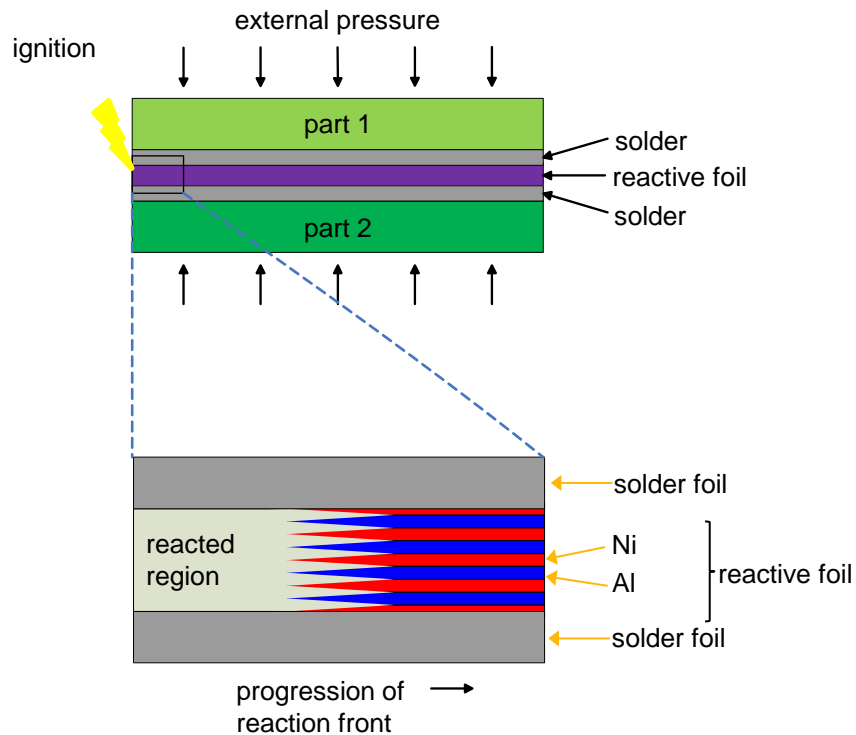
- **1960s** (esp. USSR): exothermal reactions for production of intermetallics
- **1979**, Prentice: "Heat Sources for Thermal Batteries: Exothermic Intermetallic Reactions" (US patent); **one scenario: alternating metallic layers**
- **1986**, Floro: "Propagation of explosive crystallization in thin Rh–Si multilayer films" (J. Vac. Sci. Technol. A); **preparation of nano-multilayer films**
- **1995**, Makowiecki: "Low Temperature Reactive Bonding" (patent); **films**
- **2001**, Weihs: "Method of making reactive multilayer foil and resulting product" (patent, US only); **freestanding foils**
2001, Weihs: founding of "Reactive NanoTechnologies" (now Indium Corp.); **start of commercial production of Nanofoils®**
- since then: increased usage for joining

2011, Longtin/Empa: "Benign Joining of Ultrafine Grained Aerospace Aluminum Alloys Using Nanotechnology" (Adv. Mater.)

please contact author for
further information

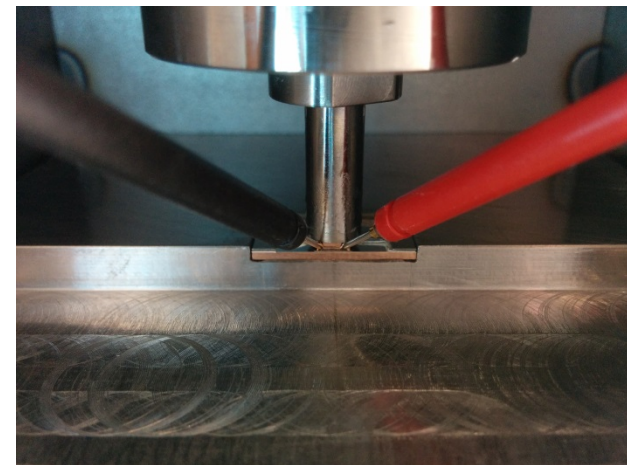
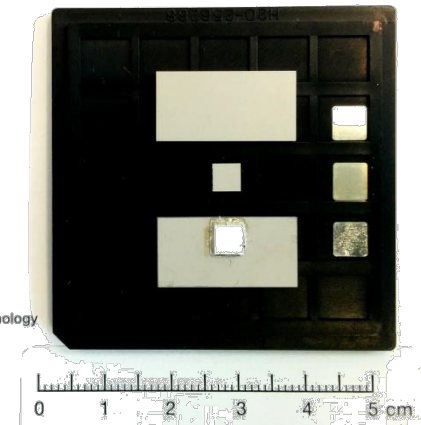
Joining with reactive nano-multilayers

Principle



Empa
Überlandstrasse 129
CH-8600 Dübendorf
Schweiz
www.empa.ch

 **EMPA**
Materials Science & Technology



alternative approach: direct
deposition of reactive nano-
multilayers (e.g. on wafer)

Joining with reactive nano-multilayers

Advantages

Processing

- localised heat source: components remain “cold”
- no furnace
- no protective atmosphere
- no flux (if clean components)
- easy handling of joining components (→pick and place)
- short processing time

Joint performance

- microstructure refinement
- good thermal properties (heat conductivity)
- stability against high temperatures & humidity

temperature-sensitive components

stress-sensitive components

controlled atmosphere, hermetic encapsulations*

prototyping

alternative to step-soldering

rework

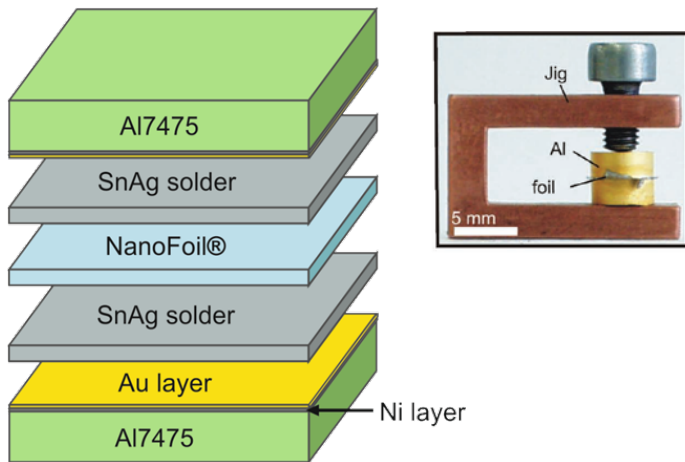
high-strength joints*

bonding of heat-sinks

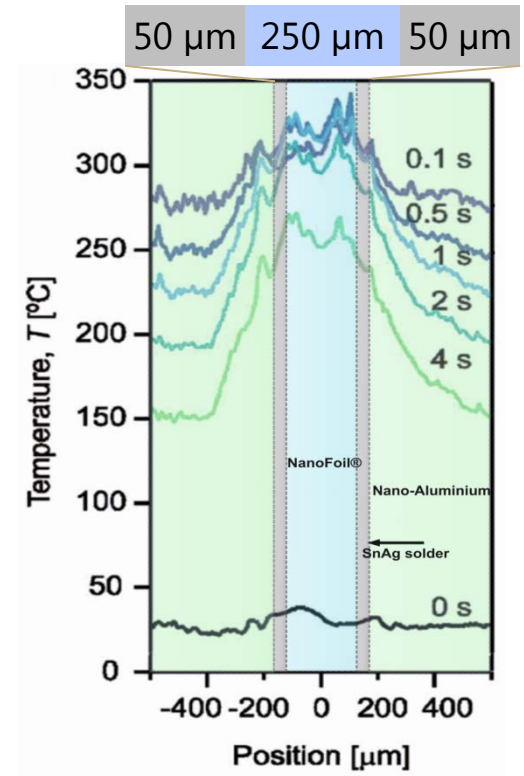
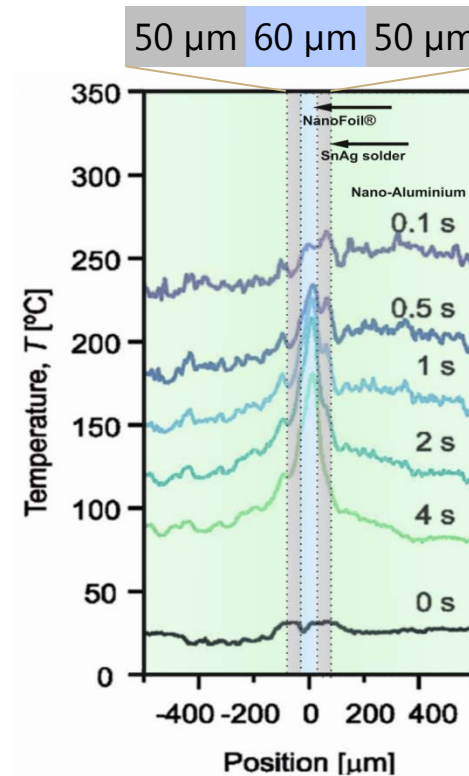
* esp. for directly deposited RNMLs

Joining with reactive nano-multilayers

Example: joining of a nano-crystalline Al alloy (Empa, 2011)



joining set-up



temperature development in joining zone

➔ **successful joining of heat-sensitive materials**

Typical problems & challenges

1. “Classical” soldering problems

- Example: Joining of stainless steel, shear strength
 - Wang 2005: **36 MPa** (Ni-Au metallisation, AgSn solder; J. Appl. Phys. 97)
 - Sen 2016: **9 MPa** (Ni metallisation, Sn + InCuSil solder, Euro Hybrid Proceedings 2016)
- ➔ **handling, cleaning, general bonding issues...**

2. Process-intrinsic problem: heat management

- no possibility for external control of process time and temperature
 - too hot: damage of components (cf. joining of nano-Al)
 - too cold: no melting of solder
 - additionally: thermo-mechanical shockwave
- ➔ **challenge: influence of substrates and components**

Joining with reactive nano-multilayers

Interreg V – Project: “Schonendes reaktives Fügen von Mikrosystemen”

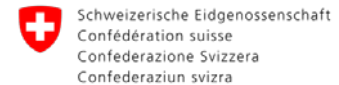
Project partner:

- Hahn-Schickard, Baden-Württemberg, Germany
- R&D in micro-assembly and packaging, sensor development,...



Project goals:

- development of *truly* benign joining processes
- characterisation of thermo-mechanical stress during joining
- design rules for reactive joining



Kanton
Zürich

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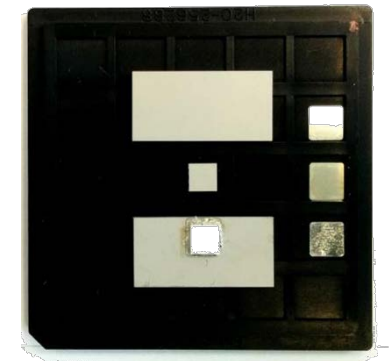
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Design of test series

- joining components
 - materials: borosilicate glass, Si, Al₂O₃, Cu
 - bond area: 4 mm x 4 mm

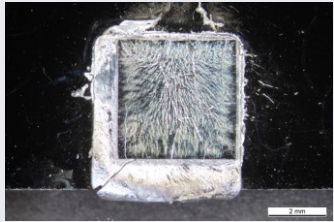
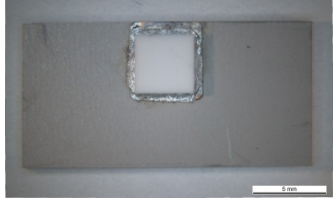
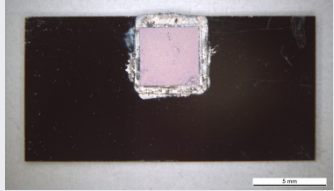

- joining setup
 - reactive system: Ni-Al, commercial nanofoils® (60 μm + 2 x 1 μm InCuSil)
 - metallisation: Ni
 - solder: Sn foils (2 x 10 μm)

- test methods
 - non-destructive (scanning acoustic microscopy, computer tomography)
 - destructive (shear strength, cross-sections)



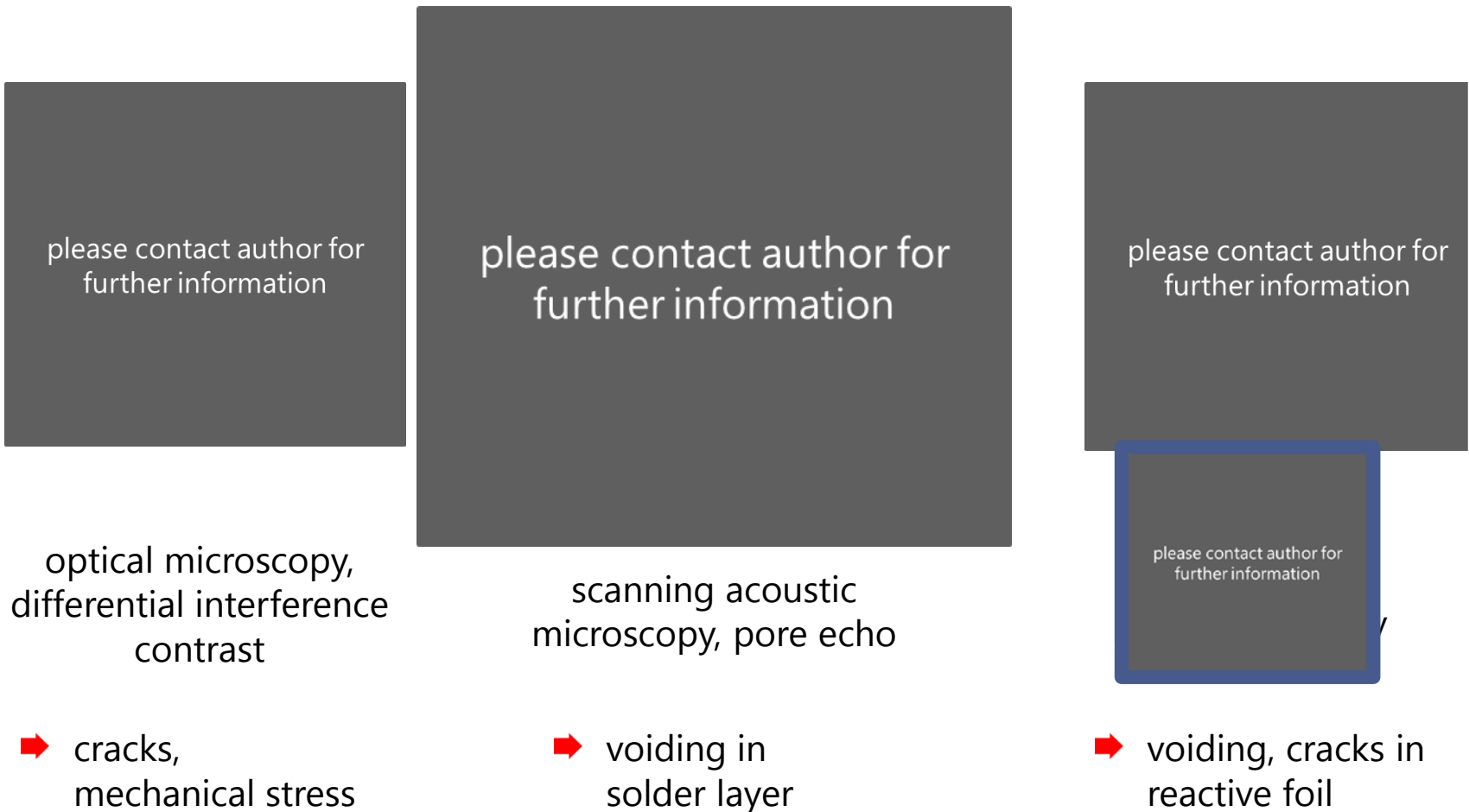
Focus: influence of substrates/components

Overview results

substrate	heat conductivity ($W \cdot m^{-1} \cdot K^{-1}$)	solderability	
borosilicate glass	1.2	joint formed, but extensive cracking	
Al_2O_3	30	good	
Si	129 (avrg.)	good	
Cu	401	no joint formed	

Focus: influence of substrates/components

Example borosilicate glass: non-destructive testing



Focus: influence of substrates/components

Example borosilicate glass: process optimisation

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optical microscopy,
differential interference
contrast

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further information

scanning acoustic
microscopy, pore echo

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further information

computer
tomography

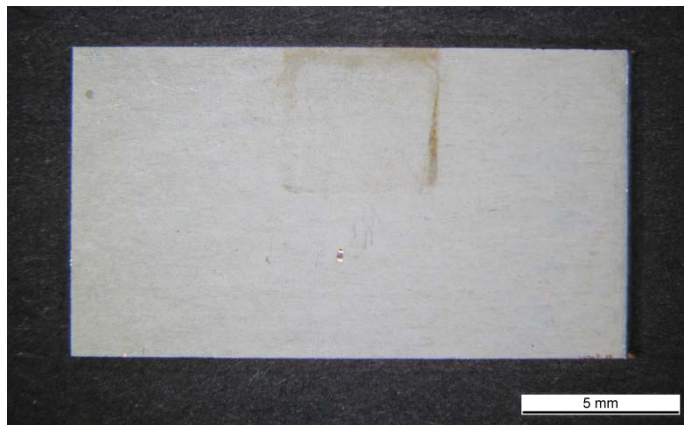
Solution: 2 x 75 μm Sn instead of 2 x 10 μm + pressure reduction

➔ no cracks in glass (but still pores)

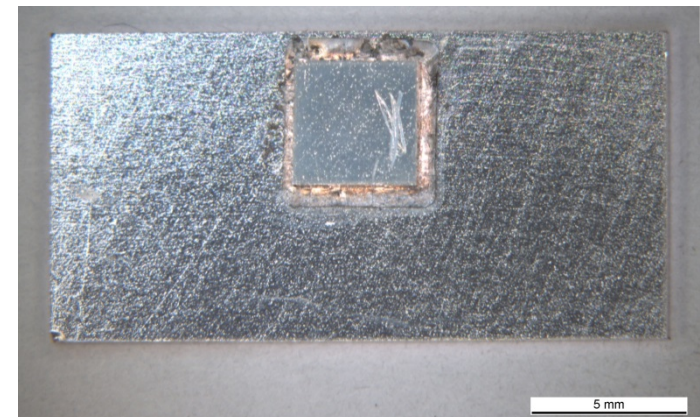
Focus: influence of substrates/components

Other materials: Al_2O_3 , Si and Cu

- Al_2O_3 and Si: some porosity, but excellent strength
 - Al_2O_3 : shear strength around 45 MPa
 - Si: fracture of substrates around 20 MPa
- Cu
 - thicker reactive foil (250 μm) = more heat generation: **unsuccessful**
 - galvanic pre-soldering of substrates: **successful**



galvanic
→
pre-tinning



Joining with RNMLs: promising new technique

- simple, fast and flexible: no furnace, no protective atmosphere, flux-free...
- benign joining possible
- hermeticity possible
- high-quality joints possible

Crucial

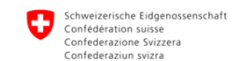
- **good soldering practice**
- **tailored joining setup for heat management:**
 - **reactive foil vs.**
 - **solder vs.**
 - **substrate/components**

Thank you for your attention!

Looking forward to your questions ...and potential cooperation projects!

Acknowledgements

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SWISS PHOTONICS

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